Application of Ring-Closing Metathesis to the Synthesis of 19-Functionalized Derivatives of 1α -Hydroxyvitamin D₃

Agnieszka Wojtkielewicz and Jacek W. Morzycki*

Institute of Chemistry, University of Bialystok, al. Pilsudskiego 11/4, 15-443 Bialystok, Poland

morzycki@uwb.edu.pl

Received November 25, 2005

ORGANIC LETTERS 2006 Vol. 8, No. 5 839-842

ABSTRACT



The synthesis of the 19-functionalized derivative of vitamin D₃ based on ring-closing metathesis (RCM) is presented.

A hormonally active metabolite of vitamin D₃, 1α ,25dihydroxyvitamin D₃, exhibits, besides the regulation of calcium and phosphorus homeostasis, a variety of biological activities such as cell differentiation and proliferation.¹ Since this discovery, extensive studies to find analogues with selective activity profiles as potential therapeutic agents have been undertaken. Over the past two decades many vitamin D derivatives containing modification in A, C, or D rings as well as in the side chain have been synthesized.² However, only a few methods of synthesis of 19-functionalized compounds have been described so far.³

10.1021/ol052856k CCC: \$33.50 © 2006 American Chemical Society Published on Web 02/01/2006

The discovery of olefin metathesis and development of well-defined ruthenium and molybdenum alkylidene catalysts⁴ has provided a very convenient synthetic route to complex olefins. The preparation of 19-functionalized derivatives of vitamin D based on cross metathesis (CM) seemed to be the shortest and most straightforward way. We have undertaken efforts to synthesize such analogues of vitamin D in our laboratory. The reactions of vitamin D₃ (1) with various olefins, such as 1-heptene, allyl alcohol, trans-3-hexenedinitrile, and allyl cyanide, were examined. Unfortunately, the CM reactions of vitamin D_3 with these alkenes in the presence of 20 mol % of Grubbs or Hoveyda second generation catalysts in dichloromethane did not work and the starting material was recovered in all cases. There was no reaction even under more drastic conditions (80 °C, 30 mol % of catalyst).

A failure of the CM approach caused alteration of the synthetic strategy and ring-closing metathesis (RCM) was used instead in the next experiments, as an alternative method for the preparation of 19-functionalized derivatives of vitamin

⁽¹⁾ Holick, M. F., Ed. Vitamin D: Physiology, Molecular Biology and Clinical Aplications; Humana Press: Totowa, NJ, 1999.

^{(2) (}a) Bouillon, R. M.; Okamura, W. H.; Norman, A. W. Endocr. Rev.
1995, 16, 200. (b) Muralidharan, K. R.; De Lera, A. R.; Isaeff, S. D.;
Norman, A. W.; Okamura, W. H. J. Org. Chem. 1993, 58, 1895. (c) Perlman,
K. L.; Sicinski, R. R.; Schnoes, H. K.; DeLuca, H. F. Tetrahedron Lett.
1990, 31, 1823. (d) Kanzler, S.; Halkes, S.; van de Velde, J. P.; Reischel,
W. Bioorg. Med. Chem. Lett. 1996, 6, 1865. (e) Kroszczynski, W.;
Morzycka, B.; Morzycki, J. W. Wiad. Chem. 2002, 56, 793.

^{(3) (}a) Yamada, S.; Nakayama, K.; Takayama, H.; Itai, A.; Iitaka, Y. J. Org. Chem. 1983, 48, 3477. (b) Yamada, S.; Suzuki, T.; Takayama, H. J. Org. Chem. 1983, 48, 3483. (c) Addo, J. K.; Ray, R. Steroids 1998, 63, 218. (d) Swamy, N.; Addo, J. K.; Ray, R. Bioorg. Med. Chem. Lett. 2000, 10, 361. (e) Addo, J. K.; Swamy, N.; Ray, R. Bioorg. Med. Chem. Lett. 2002, 12, 279. (f) Ahmed, M.; Atkinson, C. E.; Barrett, A. G. M.; Malagu, K.; Procopiou, P. A. Org. Lett. 2003, 5, 669.

^{(4) (}a) Grubbs, R. H. *Tetrahedron* **2004**, *60*, 7117. (b) Deiters, A.; Martin, S. F. *Chem. Rev.* **2004**, *104*, 2199. (c) Connon, S. J.; Blechert, S. *Angew. Chem.*, *Int. Ed.* **2003**, *42*, 1900.



D₃. According to the synthetic plan, the 3-hydroxyl group in vitamin D₃ was esterified with carboxylic acid containing the terminal double bond (4-pentenoic, 5-hexenoic, 6-heptenoic, and 10-undecenoic acid). The esterification was performed with the DCC method with DMAP as catalyst. The products 2 were obtained in high yields. The esters 2 appeared to be rather unstable and therefore were immediately subjected to RCM reaction.⁵ Although there are many reports of successful application of RCM to the synthesis of medium sized rings,⁴ in the case of metathesis of vitamin D_3 esters, the only products were dimers 3, as a result of self-methatesis (SM). Despite the change of the reaction conditions-higher temperature (80 °C), more second generation catalyst (Grubbs or Hoveyda, 30 mol %), higher dilution, and slow addition of reagents-the desired RCM products were not formed.

Compound 4 with inverted configuration at C-3 (synthesized from vitamin D_3 via the Mitsunobu reaction) did not afford the RCM product either. The same results were obtained in the 5,6-*trans*-vitamin D_3 series. The 5,6-*trans*-vitamin D_3 did not undergo the CM reactions, while the corresponding esters (5) yielded the SM products only.

Presumably, the failure of ring closure is due to the lack of sufficient conformational flexibility in ring A, as well as the steric hindrance (5,6-*cis* compounds) caused by the CD ring fragment of the molecule. Additionally, molecular modeling reveals that the cyclization requires the change of A ring conformation from the favored chair to the boat conformation.

In the next attempts, the same approach was applied for 1α -hydroxyvitamin D₃ derivatives. The 3-TBS-ether of 1α -hydroxy-5,6-*trans*-vitamin D₃ (6) obtained by a known method⁶ was analogously transformed into 3-butenoate 7, which theoretically allows for closure of a six-membered ring in the RCM reaction. In this case the desired cyclic product 8 was obtained in 70% yield with use of 20 mol % of Hoveyda second generation catalyst (worked slightly better than the Grubbs catalyst): dilution, Cm = 0.5–1.5 mM, 80 °C.⁵ The most important factor for successful RCM proved to be reaction temperature. At 40–70 °C the main product appeared to be a dimer 9, as a result of SM. The higher temperature favored the RCM reaction. Such an effect of

⁽⁵⁾ General procedure for RCM reaction: To the solution (2 mM) of 20 mol % of Grubbs (or Hoveyda) second generation catalyst in dry toluene in oven-dried Schlenk flask was added the solution (0.5-1.5 mM) of vitamin D₃ (or 1 α -hydroxyvitamin D₃) ester in dry toluene dropwise over 2 h. The reaction mixture was stirred at 80 °C for 4 h under argon atmosphere. Then the reaction mixture was concentrated in vacuo and purified by silica gel column chromatography.

⁽⁶⁾ Marshall, J. A.; Grote, J.; Shearer B. J. Org. Chem. 1986, 51, 1635.



temperature was not reported earlier for phoshonate or silic analogues of 1α -hydroxy-5,6-*trans*-vitamin D₂.^{3f} The influence of other reaction conditions (type of catalyst, dilution, mode of reagent addition) proved much less important. The RCM reaction was not observed (even at higher temperature) for 3-butenoate of 1α -hydroxyvitamin D derivatives (e.g. for compound **11**). The reason for that is presumably the greater steric hindrance caused by the CD ring fragment of the molecule because of the *cis* configuration of the 5,6 double bond in this compound.

The lactone **8** appeared to be rather unstable, and to avoid its decomposition, it was converted into the corresponding diol **12**. Subsequent photoisomerization of the 19-hydroxyethyl derivative of 1α -hydroxy-5,6-*trans*-vitamin D₃ (**12**) in



the presence of anthracene as triplet sensitizer⁷ in degassed and dry benzene at 0-5 °C afforded the desired *cis* isomer **13**. The almost complete conversion to vitamin D₃ analogue was observed within 15 min (5 × 3 min) of irradiation with a UV lamp. The two isomers (*5E* and *5Z*) were distinguished on the basis of their ¹H NMR spectra. The position of two doublets of 6-H and 7-H was diagnostic: for the 5,6-*trans* isomer these signals appeared at 6.46 and 5.89 ppm, while in the case of the 5,6-*cis* compound the first signal was shifted upfield to 6.30 ppm, the second one downfield to 5.91 ppm. Thus the distance between the doublets of 6-H and 7-H in the *trans* isomer is about 0.2 ppm larger than that between analogous signals for the *cis* isomer, what is typical for the *cis/trans* series of vitamin D derivatives.⁸ In conclusion, a new synthetic strategy for the preparation of 19-functionalized derivatives of 1 α -hydroxyvitamin D was elaborated. In the methodology developed, the synthesis of a cyclic analogue proceeds through RCM of the 1 α -hydroxy*trans*-vitamin D₃ vinylacetyl derivative. Subsequent reduction of the obtained RCM product followed by photochemical isomerization afforded the 19-functionalized analogue.

Acknowledgment. This work was supported by the State Committee for Scientific Research (Grant No. 3 T09A 013 27). We also thank Professor A. Kutner and Dr. M. Chodyński from the Pharmaceutical Institute in Warsaw for a gift of some vitamin D derivatives.

Supporting Information Available: Details of the experiments and characterization of compounds **7**, **8**, **12**, and **13**. This material is available free of charge via the Internet at http://pubs.acs.org.

OL052856K

⁽⁷⁾ Gielen, J. W. J.; Koolstra, R. B.; Jacobs, H. J. C.; Havinga, E. J. R. Neth. Chem. **1980**, *99*, 306.

⁽⁸⁾ Paaren, H. E.; DeLuca, H. F.; Schnoes, H. K. J. Org. Chem. 1980, 45, 3253.